Discussion. The refined positional parameters are in Table $1^{*}$ and bond distances and angles in Table 2. A view of the molecule along $\mathbf{b}$ is shown in Fig. 1.

The aromatic ring is, as usual, planar. The ring $\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(11)-\mathrm{O}(2)$ has a boat conformation. In the four-membered ring $C(8)-C(9)-$ $\mathrm{C}(10)-\mathrm{C}(11)$, only one of the angles deviates from $90^{\circ}$ by more than $2 \sigma$. The ring junction H atoms, $\mathrm{H}(8)$ and $\mathrm{H}(11)$, are cis. The stereochemistry is thus similar to that previously assigned to the photocycloaddition

[^0]products of substituted ethylenes to chromones (Hanifin \& Cohen, 1969).

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## 4'-Nitro-2-biphenylamine

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#### Abstract

C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{2}, M_{r}=214 \cdot 2\), monoclinic, $C 2 / c$, $a=20.66(5), \quad b=7.86(3), \quad c=13.29(1) \AA, \quad \beta=$ $104.6(2)^{\circ}, \quad V=2088 \AA^{3}, \quad Z=8, \quad D_{m}=1.37, \quad D_{x}=$ $1.36 \mathrm{Mg} \mathrm{m}^{-3}, \mathrm{Cu} K \alpha, \lambda=1.5418 \AA, \mu=0.69 \mathrm{~mm}^{-1}$, $F(000)=896, T=290 \mathrm{~K}, \quad R=0.102$ for $1175 \mathrm{ob}-$ served reflexions. The average $\mathrm{C}-\mathrm{C}$ bond in the phenyl rings is $1.391 \AA$. The molecule is non-planar; the angle between the phenyl rings is $54.7(6)^{\circ}$, the angle between the nitro group and the ring to which it is bonded is $7 \cdot 1(6)^{\circ}$ and the bond $\mathrm{N}-\mathrm{C}\left(4^{\prime}\right)$ is 1.453 (9) $\AA$.

Introduction. The structure determination of $4^{\prime}$-nitro-2-biphenylamine forms part of an investigation into liquid-crystal compounds and their chemical precursors.


Experimental. $D_{m}$ measured by flotation in aqueous cadmium $n$-dodecatungstoborate; orange opaque crystals, $0.6 \times 0.02 \times 0.05 \mathrm{~mm}$ and $0.45 \times 0.016 \times$ 0.004 mm , respectively, for $b$-axis, $k=0-6$ and $c$-axis $h k 0$ Weissenberg data. 1221 reflexions measured from multiple-film photographs, $\mathrm{Cu} K \alpha$ radiation: $-26 \leq$ $h \leq 24 ; 0 \leq k \leq 6 ; 0 \leq l \leq 16$. Data merged to give 1175 unique observed reflexions; $R_{\mathrm{int}}=0.06$; structure solved by direct methods with SHELX76 (Sheldrick, 1976) and refined by least squares ( $F$ values) with anisotropic thermal parameters for non-hydrogen atoms. H-atom positions, initially obtained from dif-
ference synthesis and placed at geometrically reasonable positions, refined with isotropic thermal parameters; $R=0.102$; unit weights. $(\Delta / \sigma)_{\max }$ in final refinement cycle 0.03 for positional and 0.04 for thermal parameters. Max. and min. heights in final $\Delta \rho$ map +0.30 and $-0.40 \mathrm{e} \AA^{-3}$. Scattering factors from International Tables for X-ray Crystallography (1974). Computer programs used: SHELX76 (Sheldrick, 1976) and locally written programs supplied by HHS and Drs C. Morgan and M. J. Mottram.

Discussion. Table 1* gives atomic parameters and Table 2 bond lengths and angles. The atomic numbering is shown in Fig. 1.

The phenyl ring with the attached amino group is planar to within $\pm 0.004 \AA$; in the other phenyl ring deviations of up to $\pm 0.01 \AA$ from the mean plane occur. The average $\mathrm{C}-\mathrm{C}$ bond in the two rings is $1.391 \AA$. The $C(1)-C(7)$ bond length, 1.465 (9) A, although slightly shorter than in some biphenyl com-

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pounds is not significantly shorter than the theoretical value of $1.477 \AA$. The $C(4)-N(1)$ bond, 1.453 (9) $\AA$, is shorter than the value 1.475 (4) $\AA$ in 4 -acetyl- $2^{\prime}$ nitrobiphenyl (Sutherland, Hogg \& Williams, 1974) and somewhat shorter than the 1.468 (9) $\AA$ in $2^{\prime}$-nitro-4-biphenylamine, but is very similar to 1.457 (2) $\AA$ in 7-nitro-2-fluorenylamine (Fallon \& Ammon, 1974). The length appears to be related to the rotation, $\varphi_{4}$, about the $\mathrm{C}-\mathrm{N}$ bond. In 4 -acetyl-2'-nitrobiphenyl and 2'-nitro-4-biphenylamine, the angle of rotation, due to steric effects, is 45.6 and $34.6^{\circ}$, respectively, whereas in the present structure it is $7.1(6)^{\circ}$ and in 7 -nitro-2-fluorenylamine it is $5.7^{\circ}$. The $\mathrm{N}(1)-\mathrm{O}(1)$ and $N(2)-O(2)$ bonds of 1.224 (8) and 1.231 (7) $\AA$ are comparable with the values quoted in other nitrosubstituted compounds (Sakore \& Pant, 1966; Sakore, Tavale \& Pant, 1967).

Table 1. Fractional atomic coordinates $\left(\times 10^{4}\right)$ and equivalent isotropic thermal parameters $\left(\AA^{2} \times 10^{3}\right)$ with e.s.d.'s in parentheses

|  | $x$ | $y$ | $z$ | $U_{\mathrm{cq}}{ }^{*}$ |
| :--- | :---: | :---: | :---: | :---: |
|  | $x$ | $3602(9)$ | $3640(4)$ | $37(4)$ |
| $\mathrm{C}(1)$ | $4193(3)$ | $4010(9)$ | $3829(5)$ | $47(4)$ |
| $\mathrm{C}(2)$ | $4876(3)$ | $2760(10)$ | $3941(5)$ | $43(4)$ |
| $\mathrm{C}(3)$ | $5360(3)$ | $1078(9)$ | $3852(5)$ | $40(4)$ |
| $\mathrm{C}(4)$ | $5156(3)$ | $612(8)$ | $3651(5)$ | $42(4)$ |
| $\mathrm{C}(5)$ | $4491(3)$ | $1886(10)$ | $3566(5)$ | $46(4)$ |
| $\mathrm{C}(6)$ | $4019(3)$ | $4956(9)$ | $3521(5)$ | $43(4)$ |
| $\mathrm{C}(7)$ | $3693(3)$ | $6224(10)$ | $2765(5)$ | $49(4)$ |
| $\mathrm{C}(8)$ | $3675(3)$ | $7512(10)$ | $2587(6)$ | $58(5)$ |
| $\mathrm{C}(9)$ | $3208(4)$ | $7552(11)$ | $3167(6)$ | $60(5)$ |
| $\mathrm{C}(10)$ | $2739(4)$ | $6335(11)$ | $3918(6)$ | $59(5)$ |
| $\mathrm{C}(11)$ | $2730(3)$ | $5015(10)$ | $4093(5)$ | $49(4)$ |
| $\mathrm{C}(12)$ | $3212(3)$ | 5010 |  |  |
| $\mathrm{~N}(1)$ | $5655(3)$ | $-259(10)$ | $3972(4)$ | $52(4)$ |
| $\mathrm{N}(2)$ | $3192(3)$ | $3798(10)$ | $4849(5)$ | $79(5)$ |
| $\mathrm{O}(1)$ | $5471(3)$ | $-1737(8)$ | $3811(4)$ | $66(4)$ |
| $\mathrm{O}(2)$ | $6249(3)$ | $142(8)$ | $4257(5)$ | $74(4)$ |
|  |  | $* U_{\text {eq }}=\left(U_{11} U_{22} U_{33}\right)^{1 / 3}$. |  |  |

Table 2. Bond lengths $(\AA)$ and bond angles $\left({ }^{\circ}\right)$ with e.s.d.'s in parentheses

| C(1)-C(2) | $1.407(8)$ | $\mathrm{C}(1)-\mathrm{C}(7)$ | $1.465(9)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.382(10)$ | $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.410(9)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.383(10)$ | $\mathrm{C}(8)-\mathrm{C}(9)$ | $1.377(10)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.383(9)$ | $\mathrm{C}(9)-\mathrm{C}(10)$ | $1.381(10)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.382(9)$ | $\mathrm{C}(10)-\mathrm{C}(11)$ | $1.386(11)$ |
| $\mathrm{C}(4)-\mathrm{N}(1)$ | $1.453(9)$ | $\mathrm{C}(11)-\mathrm{C}(12)$ | $1.416(10)$ |
| $\mathrm{N}(1)-\mathrm{O}(1)$ | $1.224(8)$ | $\mathrm{C}(12)-\mathrm{C}(7)$ | $1.395(8)$ |
| $\mathrm{N}(1)-\mathrm{O}(2)$ | $1.231(7)$ | $\mathrm{C}(12)-\mathrm{N}(2)$ | $1.396(9)$ |
| $\mathrm{C}(1)-\mathrm{C}(6)$ | $1.392(10)$ |  |  |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $121.5(7)$ | $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | $122.3(6)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | $118.3(6)$ | $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | $118.7(7)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | $122.4(6)$ | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | $121.7(7)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | $118.0(6)$ | $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | $119.2(7)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(1)$ | $122 \cdot 1(6)$ | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(7)$ | $120.2(7)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{N}(1)$ | $119.3(6)$ | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{N}(2)$ | $118.3(6)$ |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{N}(1)$ | $118.2(6)$ | $\mathrm{C}(7)-\mathrm{C}(12)-\mathrm{N}(2)$ | $121.5(6)$ |
| $\mathrm{C}(4)-\mathrm{N}(1)-\mathrm{O}(1)$ | $119.1(6)$ | $\mathrm{C}(12)-\mathrm{C}(7)-\mathrm{C}(8)$ | $118.0(6)$ |
| $\mathrm{C}(4)-\mathrm{N}(1)-\mathrm{O}(2)$ | $118.3(7)$ | $\mathrm{O}(1)-\mathrm{N}(1)-\mathrm{O}(2)$ | $122.6(7)$ |
| $\mathrm{C}(1) \mathrm{C}(7)-\mathrm{C}(12)$ | $123.1(6)$ | $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(7)$ | $120.2(6)$ |
| $\mathrm{C}(1)-\mathrm{C}(7)-\mathrm{C}(8)$ | $118.8(6)$ | $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{C}(7)$ | $122.2(6)$ |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(6)$ | $117.6(6)$ |  |  |



Fig. 1. The arrangement of the molecules in the unit cell viewed along $b$.

The planarity of biphenyl derivatives can be distorted by steric interaction of crystal packing. The angle between the rings about the central $\mathrm{C}-\mathrm{C}$ bond, $\varphi_{1}$, of $54.7(6)^{\circ}$ is smaller than the $57.1^{\circ}$ in $2^{\prime}$-nitro-4biphenylamine, but larger than the $45.9(2)^{\circ}$ in 4 -acetyl-2'-nitrobiphenyl. It is larger than that observed in $2^{\prime}$-halo-biphenyl derivatives and can be compared with $50.5^{\circ}$ in 4 -acetyl- $2^{\prime}$-fluorobiphenyl (Young, Tollin \& Sutherland, 1968) and $51.3^{\circ}$ in $2^{\prime}$-iodo-4-biphenylcarboxylic acid (Sutherland, 1970). The axis of the molecule deviates from collinearity. Not only is there a rotation $\varphi_{1}$ about the $\mathrm{C}(1)-\mathrm{C}(7)$ bond but also $\varphi_{2}$, the angle of rotation of the ring $\mathrm{C}(1)-\mathrm{C}(6)$ about an axis in its plane through $\mathrm{C}(1)$ perpendicular to $\mathrm{C}(1)-\mathrm{C}(7)$ is $0.7(6)^{\circ}$ and $\varphi_{3}=1.6(6)^{\circ}$, the corresponding angle of rotation for ring $\mathrm{C}(7)-\mathrm{C}(12)$; there is a corresponding tilt, $\varphi_{5}$, of the nitro group about $\mathrm{C}(4)$ of $0.9(5)^{\circ}$.

The shortest intermolecular contacts are $3.260(10) \AA$ between $\mathrm{C}(5)$ and $\mathrm{N}(1), 3 \cdot 310(8) \AA$ between $\mathrm{N}(2)$ and $\mathrm{O}(1), 3.413$ (10) $\AA$ between $\mathrm{N}(2)$ and $O(2)$ and $3.381(8) \AA$ between $C(6)$ and $O(1)$, where $\mathrm{N}(1), \mathrm{O}(1)$ and $\mathrm{O}(2)$ are at $1-x,-y, 1-z$ and 3.402 (10) $\AA$ between $\mathrm{C}(2)$ and $\mathrm{C}(2)$ at $1-x, 1-y$, $1-z$.

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[^0]:    * Lists of structure factors, anisotropic thermal parameters, H -atom parameters, bond lengths involving H and mean-planes' calculations have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42619 (11 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

[^1]:    * Lists of structure factors, anisotropic thermal parameters, $\mathbf{H}$ atom parameters, intermolecular contact distances and mean-plane calculations have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42634 (20 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

